Quantum Mechanics: A Pedagogical Approach to a Geometric Theory

Guilherme Dias Vianna

Dead Physicists Society

Guilherme Dias Vianna Quantum Mechanics: A Pedagogical Approach to a Geome

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QM as a closed theory The position operator The Hydrogen Atom Failures of Classical Mechanics

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QM as a closed theory The position operator The Hydrogen Atom

Failures of Classical Mechanics

One big failure

• Double Slit Experiment:



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• Double Slit Experiment:



• Particles behave like waves? Superposition effects??

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Ok, now that's bad

• The stability of the atom



• According to Larmor's formula $\left(P = \frac{q^2 a^2}{6\pi\epsilon_0 c^3}\right)$

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- According to Larmor's formula $\left(P = \frac{q^2 a^2}{6\pi\epsilon_0 c^3}\right)$
- How come matter is stable?

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So...how to fix this?

• What else did we found that was weird??

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Failures of Classical Mechanics

So...how to fix this?

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 - Most physical quantities (energy, for instance) take discrete values in the atomic scale!

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 - Only certain "configurations"seem right!

The Postulates Applications

Quantum Mechanics!

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The Postulates Applications

Each more beautiful than the last!

First Postulate:

All the information of a physical system (or, more precisely, it's state) is represented by a vector $|\phi\rangle$ in a given Hilbert space \mathscr{H} .

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The Postulates Applications

A small detour in eigenproblems

The eigenvalues of an operator $\hat{\mathcal{A}}$ are the numbers that satisfy:

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If the dimension is **infinite**, we need to solve a **differential** equation!

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Third Postulate: The **probability** of measuring certain observable O with value v_o is given by:

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Here, $\langle \phi | \psi \rangle \longleftarrow \phi \cdot \psi$ Imposing that $\|\psi\| = \|\phi\| = 1$, we get that $\langle \phi | \psi \rangle$ is the **projection** of $|\psi\rangle$ onto the direction of ϕ !

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The Postulates Applications

THE equation

Fourth Postulate: The system evolves in time according to

$$i\hbar \frac{\mathrm{d}|\psi\rangle}{\mathrm{d}t} = \hat{\mathcal{H}}|\psi\rangle$$

Schrödinger's Equation

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EXPONENTIAL OF A MATRIX??

The Postulates Applications

A second, more fun detour

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$$e^{\hat{\mathcal{A}}} = \mathbb{1} + \hat{\mathcal{A}} + \frac{1}{2}\hat{\mathcal{A}}^2 \dots$$

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The Postulates Applications

And why they are reasonable?

• The use of vectors and complex numbers guarantees superposition!

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And why they are reasonable?

- The use of vectors and complex numbers guarantees superposition!
- Hermitian operators always have real eigenvalues!
- The use of scalar products guarantees that the usual probability laws hold just fine!

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The Postulates Applications

The Ethylene Molecule

1**530 cm-1**



The Postulates Applications



The most general hamiltonian in this case is:

$$\hat{\mathcal{H}} = \begin{pmatrix} E_0 & -A \\ -A & E_0 \end{pmatrix}$$

It's eigenvalues/vectors are:

$$E_{\pm} = E_0 \mp A, \ |\chi_{\pm}\rangle = \frac{1}{\sqrt{2}}(|\phi_1\rangle \pm |\phi_2\rangle)$$

The Postulates Applications

By projecting $|\chi_{\pm}\rangle$ onto the basis of position $|x\rangle$ (more to come about this):



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Some difficulties The solution!

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Reason: Since there are a continuous number of "wanted eigenvalues" and the domain of this operator is not quite clear, this operator is **unbounded** in \mathscr{H} ! (so $|\mathbf{x}\rangle$ are not really elements of \mathscr{H} , even though they form a of basis!)

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The third and final detour

In reality, $|\mathbf{x}\rangle$ are elements of the space of tempered **distributions** \mathscr{D} ! It can be shown that $\mathscr{H} \subset \mathscr{D}$

They form a basis in the a sense and are subject to the relation:

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If we project a state vector $|\psi(t)\rangle$ onto this subspace we get something called a wavefunction!

$$\langle \mathbf{x} | \psi \rangle = \psi(\mathbf{x}, t)$$

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Schrödinger equation on position space The problem of the atom The solution

First rule of position club:

By projecting Schrödinger's equation on the subspace of position:

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We get it on the representation of space:

$$i\hbar \frac{\partial \psi}{\partial t}(\mathbf{x},t) = \mathcal{H}\psi(\mathbf{x},t)$$

A partial differential equation!

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The hamiltonian for a single particle is:

$$\mathcal{H} = -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{x})$$

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What have you brought upon this cursed land...

The hamiltonian for a single particle is:

$$\mathcal{H} = -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{x})$$

Where $\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$ is called the laplacian operator In the case of the hydrogen atom:

$$V(r) = -\frac{1}{4\pi\epsilon_0} \frac{e^2}{r}$$

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Schrödinger equation on position space The problem of the atom The solution

The eigenvalue problem

I could try to walk you through the resolution of what's left

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- We need to rewrite ∇^2 in spherical coordinates

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$$\nabla^2 = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \varphi^2}$$

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• Then we would need to solve the eigenvalue problem:

$$\left(-\frac{\hbar^2}{2m}\nabla^2 + -\frac{1}{4\pi\epsilon_0}\frac{e^2}{r}\right)\psi = E\psi$$

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• We can use the separation of variables (because that's God's command):

$$\psi(r,\theta,\varphi)=R(r)\Theta(\theta)\Phi(\varphi)$$

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"Detailed "resolution

The rest is just trivial computation and

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- Special functions

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- Lots of coefficients and people's names (Laguerre, Legendre...)

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The eigenfunctions, finally

We finally get:

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Schrödinger equation on position space The problem of the atom **The solution**

The eigenfunctions, finally

We finally get:

 $\psi_{nlm}(r,\theta,\varphi) \propto e^{-r} r^l L_{n-l-1}^{2l+1}(r) Y_m^l(\theta,\varphi)$

Schrödinger equation on position space The problem of the atom The solution

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Quantum Numbers			
n	1	m_l	Eigenfunctions
1	0	0	$\psi_{100} = \frac{1}{\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} e^{-Zr/a_0}$
2	0	0	$\psi_{200} = \frac{1}{4\sqrt{2\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \left(2 - \frac{Zr}{a_0}\right) e^{-Zr/2a_0}$
2	1	0	$\psi_{210} = \frac{1}{4\sqrt{2\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \frac{Zr}{a_0} e^{-Zr/2a_0} \cos\theta$
2	1	± 1	$\psi_{21\pm 1} = \frac{1}{8\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \frac{Zr}{a_0} e^{-Zr/2a_0} \sin \theta \ e^{\pm i\varphi}$
3	0	0	$\psi_{300} = \frac{1}{81\sqrt{3\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \left(27 - 18\frac{Zr}{a_0} + 2\frac{Z^2r^2}{a_0^2}\right) e^{-Zr/3a_0}$
3	1	0	$\psi_{310} = \frac{\sqrt{2}}{81\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \left(6 - \frac{Zr}{a_0}\right) \frac{Zr}{a_0} e^{-Zr/3a_0} \cos\theta$
3	1	± 1	$\psi_{31\pm 1} = \frac{1}{81\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \left(6 - \frac{Zr}{a_0}\right) \frac{Zr}{a_0} e^{-Zr/3a_0} \sin \theta \ e^{\pm i\varphi}$

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And does it work?

This solutions gives us:

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Do we need QM? QM as a closed theory The position operator The Hydrogen Atom

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And so...

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Quantum physics just got less complicated

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Thank you!

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